It is to be hoped that further exploration of the data will continue, and that this monograph will not constitute the authors' final words on what is to be learnt from the study they have carried out so carefully.

REFERENCES


Effect of 2,3-Dimercaptopropanol (BAL) on Urinary Excretion and Brain Content of Mercury

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2,3-Dimercaptopropanol (BAL) given to rats in doses used for the treatment of acute mercury chloride poisoning in man increases the excretion of mercury. The mercury content of the brain was unaffected by 2,3-dimercaptopropanol unless it was given a short time after the mercuric chloride, when there was a 24% increase. The relevance of these findings to therapy with BAL is discussed.

2,3-Dimercaptopropanol (BAL) treatment of acute poisoning by mercury salts is sometimes regarded as contraindicated on the grounds that in certain circumstances it did not increase mercury excretion (Fitzsimmons and Kozelka, 1950; Berlin and Lewander, 1965) and did increase the concentration of mercury in the brain (Berlin and Lewander, 1965). In this paper I show that in rats therapeutic doses (recommended by the Extra Pharmacopoeia Martindale, 1967) increased mercury excretion and, unless given very shortly after the injection of mercuric chloride, did not increase the mercury content of the brain. The previous findings are wholly explicable by the use of doses lower than those used in man and by the timing of the therapy.

Materials and Methods

Female albino rats, Porton Wistar strain, of 200-240 g. body weight, were injected, intramuscularly into one thigh, with 100 µg. mercury as mercuric chloride labelled with 203Hg to a specific activity of 3.5 or 7.5 µC/mg. (Radiochemical Centre, Amersham, Bucks). BAL (Boots Pure Drug Co.) in 0.1 ml. arachis oil was given intramuscularly into the other thigh. When the urine and faeces were collected the animals were placed in metabolism cages. The animals were killed by decapitation. The mercury content of the urine, faeces, and brain was estimated by counting 203Hg with an automatic scintillation spectrometer (Model 50A, Packard Instrument Co.) having a counting efficiency of 40%.

Results

When BAL was given to rats six days after the administration of mercuric chloride a dose of 2 mg./kg. had no effect on the urinary excretion of mercury, but doses of 8 and 16 mg./kg. markedly increased it. BAL in doses of 4 mg./kg. had a marginal effect on urinary excretion. The faecal excretion remained unchanged except after the highest dose which slightly increased it (Figure).

Table I shows that doses of 6 and 16 mg./kg. given five days after the administration of mercuric chloride increased the urinary excretion but had no effect on the mercury content of the brain in the animals killed 24 hrs after the injection of BAL.

However, when BAL (6 mg./kg.) was given 90 min. after the mercuric chloride and the treatment was repeated 5, 25, 30, and 48 hrs later there was an increase in the mercury content of the brain in the animals killed 24, 48, and 72 hrs after the first injection. Table II shows that the increase in the brain mercury content was only about 24% but the urinary excretion of mercury was 2.5 times higher in the BAL-treated than in the control animals.
Notes and Miscellanea

Discussion

Longcope, Luetscher, Calkins, Grob, Bush, and Eisenberg (1946) and Moeschlin (1964) showed that BAL treatment was life-saving in acute cases of mercuric chloride poisoning. They gave doses of 10-15 mg./kg./day, comparable with the higher doses used in my experiments. Berlin and Lewander (1965) gave only 2 mg./kg./day to mice, which is a far lower dose. At this dose rate excretion by rats is not increased (Figure). Fitzsimmons and Kozelka (1950) did not state what dose they used, but it seems probable that it was lower than that used in man. There is, therefore, no inconsistency between the published data on animals and the effectiveness of BAL therapy. [Very high doses (50 mg./kg.) have also been shown to accelerate the excretion of mercury from rats (Swensson and Ulfvarson, 1967)].

The mercury content of the brain was increased when BAL was given to the rats very shortly after the mercuric chloride (Table II), confirming Berlin and Lewander's report on mice. The increase in content was, however, only about 24%. The relative unimportance of such an increase is illustrated by the work of Berlin, Jerksell, and von Ubisch (1966), who showed that when metallic mercury was given the brain content, as a fraction of the total body burden, was 10 times greater than when a mercury salt was given.

The author wishes to thank Mr. R. Lock for technical assistance.

TABLE I
EFFECT OF A SINGLE DOSE OF BAL GIVEN 5 DAYS AFTER THE INJECTION OF 100 μG. Hg as HgCl₂ ON THE URINARY EXCRETION AND BRAIN CONTENT OF MERCURY

<table>
<thead>
<tr>
<th>Treatment</th>
<th>No.</th>
<th>Urine (24 hrs)</th>
<th>Brain</th>
</tr>
</thead>
<tbody>
<tr>
<td>None</td>
<td>14</td>
<td>0.07 (±0.07)</td>
<td>0.0520 (±0.0018)</td>
</tr>
<tr>
<td>BAL, 6 mg./kg.</td>
<td>12</td>
<td>3.20 (±0.816)</td>
<td>0.0523 (±0.0037)</td>
</tr>
<tr>
<td>BAL, 16 mg./kg.</td>
<td>12</td>
<td>14.74 (±0.864)</td>
<td>0.0504 (±0.0024)</td>
</tr>
</tbody>
</table>

(The numbers in brackets are the standard errors). The animals were killed 24 hrs after the BAL treatment.

1.0.02 < p < 0.025
2.0.0005 > p

TABLE II
URENA Excretion and Brain Content of Mercury in Rats when BAL Treatment (6-0 mg./kg. b.d.) Was Started 90 Min. after the Mercury Injection

<table>
<thead>
<tr>
<th>Time after First BAL Injection (hrs)</th>
<th>Brain Mercury Content (μg.)</th>
<th>Accumulated Urinary Excretion (μg.)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Control</td>
<td>BAL-Treated</td>
<td>Control</td>
</tr>
<tr>
<td>---------</td>
<td>-------------</td>
<td>---------</td>
</tr>
<tr>
<td>24</td>
<td>0.030</td>
<td>0.039</td>
</tr>
<tr>
<td></td>
<td>0.049</td>
<td>0.053</td>
</tr>
<tr>
<td>48</td>
<td>0.033</td>
<td>0.049</td>
</tr>
<tr>
<td></td>
<td>0.041</td>
<td>0.061</td>
</tr>
<tr>
<td>72</td>
<td>0.039</td>
<td>0.048</td>
</tr>
<tr>
<td></td>
<td>0.046</td>
<td>0.054</td>
</tr>
<tr>
<td></td>
<td>0.049</td>
<td>0.055</td>
</tr>
<tr>
<td>Average</td>
<td>S.E.M.</td>
<td>0.041</td>
</tr>
<tr>
<td></td>
<td></td>
<td>0.0025</td>
</tr>
</tbody>
</table>

REFERENCES


Notes and Miscellaneous


Stress Fracture of the Femur: A Work Accident

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Stress fracture has been well documented. Most cases reported occur in young conscripts of the armed forces (Bingham, 1945; Mann, 1945; Wolfe and Robertson, 1945). Other causes include the hard training programme of athletes, sport (Devas and Sweetnam, 1956; Kitchin, 1948; Devas, 1958; Miller, 1960; Devas, 1963), ballet dancing (Burrows, 1956), and ice-skating (Ingersoll, 1943). Presented here is a case of stress fracture of the femur caused by hard physical labour.

Case History

M.D., a casual labourer of 35 years of age, obtained employment during the construction of a building. His work entailed the continuous carrying of heavy loads from the ground to the upper floors. Three weeks after starting work he began to feel pain in the lower left thigh during work, which was at first relieved by rest. This gradually became worse and he sought medical advice on December 17, 1965. Radiographs were normal and a muscle strain was diagnosed. He continued working in order not to be discharged from the pay-roll. The pain became more persistent and severe until, by the end of a further three weeks, he could not bear weight fully on the leg, and limped. Three days before admission to hospital, on January 6, 1966, his left foot caught on the edge of a stair at home. He did not fall, but the pain suddenly became excruciating and he could bear no weight on it.

Examination showed the usual signs of fracture but without deformity. Radiographs (Fig. 1) showed a fracture with minimal displacement.

Investigation showed: HB 14.5 g. %, W.B.C. 7,700/ c.mm. (band 4%, segment 72%, monocytes 6%, lymphocytes 18%); E.S.R. 32 mm. in one hour (Westergren); urea 33 mg. %, blood sugar 118 mg. %; total serum proteins 7 g. % (albumin 4 g. %); acid phosphatase 0.3 unit, alkaline phosphatase 2.0 units; potassium 4.5 mEq/l, sodium 140 mEq/l, chlorides 100 mEq/litre.

He was given a single hip spica without anaesthesia and was discharged on January 14. His convalescence was uneventful and on March 6 he was allowed to bear weight on the plaster. On April 14 the plaster was removed. Clinically and radiologically (Fig. 2) the fracture had united.

Discussion

The insidious onset, the nature of his work, the trivial cause of the fracture, plus the lack of rotational deformity and little displacement occurring in a bone of normal texture and density, point to a diagnosis of stress fracture. Freiberger and Mayer (1964) report a case of bilateral stress fracture of the first rib in a plasterer who spent many hours plastering ceilings, holding a board of wet plaster in his left hand and a trowel in his right. This is the only other case of occupational stress fracture mentioned in the literature.

REFERENCES


