Does occupational exposure to iron promote infection?

Keith Palmer, David Coggon

Introduction
Siderosis, the accumulation of ferric oxide particles in the lung, was first described by Zenker over a century ago. Enquiries subsequently focused on the effect of this pneumoconiotic process on lung function, generally concluding that it was benign. More recently, however, it has become apparent that iron influences bacterial virulence and has a role in host defence against infection. The evidence accrued is sufficient to trigger a reappraisal of health risks in occupational groups exposed to iron. This paper reviews some of the evidence linking iron with infection, and considers what information exists on risk in the occupational setting.

IRON IN BIOLOGICAL SYSTEMS
Fifty years ago Schade and Caroline discovered that iron binding proteins, present in blood and the whites of eggs, could inhibit bacterial growth in vitro. They hypothesised that the proteins they had discovered bound iron so tightly that bacteria could not obtain enough of it to support growth—an effect they were able to abolish by adding extra iron. Later on, others showed that animals injected with iron were more susceptible to infection than untreated controls, and that the well known antibacterial effects of body fluids could be abolished in vitro by adding iron.

As Schade and Caroline suggested, a check is placed on growth of pathogens because the amount of free iron available to them in the body fluids of humans and animals is extremely limited. Most of the body's iron stores are intracellular—in ferritin, haemosiderin or haem; and the extracellular fraction is bound to high affinity iron binding proteins—transferrin in serum and lactoferrin in external secretions. These proteins have large association constants, and are only partially saturated under normal circumstances (30% to 40% in the case of serum transferrin), so the amount of free iron in equilibrium with iron binding proteins is thought to be as low as 10^-10 M.

This arrangement makes sense. The ease with which iron undergoes changes in its oxidative state by electron transfer makes it an ideal biological catalyst, essential in the life processes of prokaryotes, eukaryotes, and anaerobic, photosynthetic, and nitrogen-fixing life forms, but also a focus for potentially injurious free radical formation. Under physiological conditions ferric iron tends to oxidise, hydrolyse, and polymerise, forming relatively insoluble ferric hydroxide and oxyhydroxide polymers. The absorption, transfer, and delivery of iron is tightly controlled at every stage, to ensure that it remains available in a soluble, non-toxic form. Commensal microorganisms and microbial pathogens are therefore thought to exist in an iron restricted environment.

THE BATTLE FOR FREE IRON
There is evidence that pathogens adapt in a variety of ways to obtain the iron they need—for example, by producing their own low molecular weight iron chelators, by modifications to their outer membrane proteins, and by the elaboration of haemolysins which liberate iron from haem. Host organisms counter this by restricting iron availability during infection. Additional iron binding capacity may be recruited in inflammatory exudate, as polymorphonuclear leucocytes degranulate, releasing lactoferrin, while the amount of iron bound to serum transferrin falls (the hypoferraemia of infection) by a mechanism that may entail lactoferrin release, macrophage sequestration of Fe^-3- transferrin complexes, and increased synthesis of ferritin. Dietary iron assimilation is suppressed by as much as 80%; and iron efflux from macrophages that have digested effete red blood cells is reduced by as much as 70%.

Other interactions between organism and host may also operate, including the host's immune response to foreign iron sequestering proteins, and the proteolytic cleavage of transferrin and lactoferrin by certain bacteria. The essential point is that a critical balance exists between commensal or pathogen and host in the fight for available iron. The normal flora of the respiratory tract reflect in part the nutrient limited balance so achieved. In situations where the balance is disturbed, as for example when exogenous or endogenous supplies of iron exceed the capacity of the iron binding protein system, overgrowth of organisms may be encouraged.

Similar considerations may apply in viral infection: although viruses do not require iron, the host cells they infect need iron before viral replication can occur, and the hypoferraemia of infection has been found in children infected with mumps and chickenpox.

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<tbody>
<tr>
<td></td>
<td>Ages 20–65</td>
<td>Deaths observed</td>
<td>SMR (95% CI)</td>
<td>Deaths observed</td>
</tr>
<tr>
<td>Pneumonia</td>
<td>Metal fume (definite)†</td>
<td>285</td>
<td>167 (148 to 187)</td>
<td>198</td>
</tr>
<tr>
<td></td>
<td>Metal fume (possible)‡</td>
<td>254</td>
<td>158 (139 to 179)</td>
<td>139</td>
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<tr>
<td>Influenza</td>
<td>Metal dust §</td>
<td>111</td>
<td>174 (143 to 209)</td>
<td>34</td>
</tr>
<tr>
<td></td>
<td>Metal fume (definite)†</td>
<td>101</td>
<td>133 (108 to 161)</td>
<td>69</td>
</tr>
<tr>
<td></td>
<td>Metal dust (possible)‡</td>
<td>77</td>
<td>131 (103 to 163)</td>
<td>58</td>
</tr>
<tr>
<td>Bronchitis**</td>
<td>Metal fume (definite)†</td>
<td>24</td>
<td>103 (66 to 153)</td>
<td>18</td>
</tr>
<tr>
<td></td>
<td>Metal fume (possible)‡</td>
<td>118</td>
<td>173 (124 to 207)</td>
<td>384</td>
</tr>
<tr>
<td></td>
<td>Metal dust §</td>
<td>83</td>
<td>115 (92 to 142)</td>
<td>289</td>
</tr>
</tbody>
</table>

The data are derived from the Registrar General's decennial supplements, England and Wales for the relevant periods. For 1979–80 and 1982–90 the PMRs are standardised for both age and social class; for earlier periods the PMRs and SMRs are standardised for age, and adjusted for social class by multiplying expected numbers of deaths by the relevant social class specific PMR or SMR for men of the relevant age group.


Other mechanisms of action
Apart from its role as a nutrient for pathogens, there are other possible ways in which iron could promote infection, especially respiratory tract infection. Iron may be implicated through a mechanism of free radical injury: in vivo iron dependent reduction of hydrogen peroxide generates hydroxyl radicals, the toxic properties of which have recently been reviewed. Studies have shown that metal particles, or carbon coated with metals, can be cytotoxic to macrophages, and short term inhalation experiments in animals have produced a cytotoxic response at ambient concentrations down to 0.1 mg/m³. Factors that interfere with the efficiency of phagocytosis are liable to render the host more susceptible to infection, independent of any effect on the nutrient status of the pathogen.

Excess iron in vivo
Observations on iron overload in vivo are generally consistent with the in vitro experimental data. In clinical situations where the availability of free iron is increased, a propensity to infection has been described—for example, patients with sickle cell disease who release free iron in haemolytic crises, seem to be more susceptible to infection, particularly pneumonia and pneumococcal infection, patients with haemochromatosis, who absorb excessive quantities of iron from dietary sources are prone to infection with Vibrio species, accidental iron overdose has been linked with bacterial septicaemia and meningitis, low concentrations of unsaturated transferrin have been associated with high fatality in pneumococcal pneumonia, and lactoferrin deficiency has been described in the polymorph granules of a patient with repeated deep seated abscesses. These and other clinical consequences of excess iron have been reviewed recently.

Occupational exposure to iron
Sources of exposure
Occupational exposure to iron arises mainly from work that generates metal fume or metal dust. Occupations associated with metal fume include gas and electric welders, cutters and braziers; furnacemen in foundries and iron and steel production; and foundry moulders and core makers, who in smaller foundries also pour and cast molten metals. Exposure to metal fume may also occur, although to a lesser extent, in foundry and steel-mill labourers, and in sheet metal workers, who sometimes cut metal sheets with welding apparatus. Occupations that may incur exposure to iron dust include: fettlers, metal polishers, boiler scalers, and workers engaged in the mining, crushing, milling, and mixing of iron ores.

The occurrence of siderosis in several of these groups, such as welders, fettlers, dressers, boiler scalers, and iron ore miners' attests to considerable degrees of iron exposure. In 1955–60 the prevalence of siderosis among welders and burners in the fettling and grinding shops of a Sheffield foundry was found to be 17.6%; others too have reported its frequent occurrence.

Commonly the occupational exposure is to ferric oxide, although in the case of welding the fume is more complex (20%–90%) is crystalline, especially as Fe₂O₃, but it also contains other compounds, including fluorides of sodium and calcium, carbonates of sodium and potassium, magnesium oxide, and Mn₃Fe₂O₆. In metal mining the main ores are haematite and magnetite.

Lung responses to iron overload
The lung's response to excessive iron may include the liberation of iron binding protein, extra macrophage uptake, and ferritin formation. Interestingly, simulated welding exercises provoke a considerable increase in the polymorphonuclear count in bronchoalveolar lavage fluid, together with a release of cytokines, including TNF, IL-1, and IL-6. Perhaps this inflammatory cellular response leads to the release of lactoferrin and the sequestration of free iron, as the IL-1 response in infection has been linked with the hypoferraemia of infection.

In any case, the response is not always complete or adequate. In siderosis ferric oxide particles have been found extravasally, in the
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### Table 2: Mortality from non respiratory infectious diseases among men in metal working occupations in England and Wales, 1979-80 and 1982-90

<table>
<thead>
<tr>
<th>Underlying cause of death*</th>
<th>All non-respiratory infections</th>
<th>Viral infections other than influenza (045-57, 060-6, 070-9)</th>
<th>CNS infections (320-4)</th>
<th>Infections of the kidney; cystitis (510,995)</th>
<th>Septicaemia (038)</th>
<th>Cancer of the lung and pleura (162):</th>
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<tbody>
<tr>
<td></td>
<td>W/SMR (95% CI)</td>
<td>Deaths PMR (95% CI)</td>
<td>Deaths PMR (95% CI)</td>
<td>Deaths PMR (95% CI)</td>
<td>Deaths PMR (95% CI)</td>
<td></td>
</tr>
<tr>
<td>20-64</td>
<td>37 98 (69 to 136)</td>
<td>26 93 (60 to 136)</td>
<td>24 109 (70 to 162)</td>
<td>16 81 (46 to 131)</td>
<td></td>
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<tr>
<td>65-74</td>
<td>37 98 (69 to 136)</td>
<td>26 93 (60 to 136)</td>
<td>24 109 (70 to 162)</td>
<td>16 81 (46 to 131)</td>
<td></td>
<td></td>
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</tbody>
</table>

† Welders, moulders and coremakers, and foundry furnace men.
‡ Foundry labourers, sheet metal workers, tin platers, and galvanisers.

alveolar walls and spaces. Some of the iron is stored as ferritin, and so stains with Perl's Prussian blue, but much of it does not. This confirms a situation of iron overload. It should be noted in this context that even a partial increase in iron saturation may make iron more freely available to some well adapted pathogens.

What factors are likely to determine the type and magnitude of effect? As in other occupational exposures, dose is important, but the relative importance of intensity and duration of exposure is not clear. Perhaps brief peaks of exposure matter in relation to short term adaptive responses in the host, whereas the outcome of long term exposure depends upon the lung's adaptive reserves.

Particle size may influence local dose and delivery site. Two different methods of particle generation are in operation occupationally—condensation of heated metal fume to produce very fine particulate metal oxides, and abrasion and dispersion of iron bearing materials to produce somewhat larger iron laden dusts. It might be supposed that these mechanisms result in different patterns of deposition and different clinical effects. In practice, however, there is overlap: metal oxide condensates in metal fume tend to aggregate and grow and to deposit in large as well as small airways, whereas the occurrence of siderosis and alveolar iron in the dust generating professions confirms that these activities can also produce considerable numbers of fine particles.

The effect may also depend on host susceptibility and a multiplicity of host defence factors—some related to iron (including differences in iron body stores and iron binding capacity) and some not. Constitutional factors, nutrition, pre-existing disease, and cigarette and alcohol intake are all likely to play a part. Finally, the outcome may depend upon the resident microbial population. For example, workers with chronic bronchitis who have a permanent reservoir of pathogens in their bronchi may be at different risk from other healthier workers.

To date interest has mainly focused on the role of endogenous and dietary iron in infection, but some occupational data exist that provide information on the relation between exogenous respirable iron and infective illness.

### Epidemiology

**Occupational Mortality**

Direct evidence that occupational exposure to iron increases susceptibility to infection comes mainly from routinely published analyses of occupational mortality. Coggon et al. recently conducted an analysis of mortality from pneumonia in populations exposed to metal fumes, with data abstracted from the Registrar General...

In the decennial supplements, expected numbers of deaths for each occupation in men of working age were calculated by applying five year age specific death rates for pneumonia in the general population to an estimate of the occupational population derived from the national census in the middle year of the study period (1961 or 1971). The relation of observed to expected deaths was expressed as a standardized mortality ratio (SMR). The report for 1970–2 also presented proportional mortality ratios (PMRs) for the age range 65–74 years, with expected numbers of deaths derived by applying age specific proportions of death from pneumonia in the general population to the total number of deaths at each age in the occupational population. Death reports for 1979–80 and 1982–90 were used to derive PMRs for pneumonia by occupation, standardised for age in five year strata and for social class (SMRs could not be calculated as the population denominators for this period were not available).

The analysis showed a consistently increased mortality from pneumonia in occupations involving exposure to metal fume, and particularly in welders, moulders, and coremakers. The largest excess of deaths was for lobar pneumonia, but increases were also found for other subcategories with the exception of bronchopneumonia. Moreover, the higher risk was restricted to men below retirement age, suggesting a short term, reversible effect of exposure.

Table 1 presents these observations with added data derived from the Registrar General's supplements for 1930–219 and 1949–53.20 Only SMRs were published for this period. In table 1 we have grouped occupations with potential exposure to metal fume into those in which exposure is definite (welders, moulders and coremakers, and foundry furnacemen), and those where it is possible (foundry labourers, sheet metal workers, tin platers, and galvanisers); included a group of workers with definite metal dust exposure (metal polishers and fettlers); and examined influenza and bronchitis as other causes of death that might also be influenced by iron exposure. The data for 1979–80 and 1981–90 have been standardised for age and social class, and those for earlier periods standardised for age, and adjusted for social class by multiplying the expected numbers of deaths by the corresponding cause and social class specific SMR for men of the relevant age group.

As in previous analyses, occupations with potential exposure to metal fume, and particularly welders, moulders, coremakers, and furnace men, had significantly increased mortality from pneumonia. Excesses were also apparent for metal grinders and polishers, although they were less obvious. The pattern for influenza (deaths from which may result from secondary bacterial pneumonia21 22) was broadly similar, although by the 1980s there were too few deaths observed and expected for meaningful interpretation. Occupations entailing exposure to metal fume and dust also tended to have increased mortality from bronchitis and again the highest rates were in welders, moulders, coremakers, and furnace men.

As previously noted, the excess mortality from pneumonia in occupations with exposure to metal fume has been confined to men of working age. This makes confounding by smoking and other non-occupational factors an unlikely explanation. In further support of this, the occupations do not have comparable excesses of lung cancer (table 2), nor is their mortality from non-respiratory infections increased.

These findings support the hypothesis that inhalation of metal fume and dust promotes respiratory infection, but they do not indicate whether the effect is specific to iron. A test of this is provided by data from 1930–2 and 1949–53 when (by contrast with later reports) workers in ferrous foundries were distinguished from furnace men working other metals, and iron ore miners from miners of tin, copper and other metallic ores (table 3). The increase of mortality from pneumonia and influenza was generally more obvious in those working with ferrous metal, but no consistent differences were apparent in mortality from bronchitis. Similarly, in 1910–2 the risk of
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### Pneumonia

<table>
<thead>
<tr>
<th>Years</th>
<th>Age Group</th>
<th>Observed Deaths</th>
<th>SMR (95% CI)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1930-2</td>
<td>20-65</td>
<td>158 (223)</td>
<td>191 (162-233)</td>
</tr>
<tr>
<td>1949-53</td>
<td>20-64</td>
<td>22 (108-261)</td>
<td>173 (150-195)</td>
</tr>
<tr>
<td>1930-2</td>
<td>20-65</td>
<td>237 (317)</td>
<td>213 (244)</td>
</tr>
<tr>
<td>1949-53</td>
<td>20-64</td>
<td>20 (302)</td>
<td>200 (122-309)</td>
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<thead>
<tr>
<th>Years</th>
<th>Age Group</th>
<th>Observed Deaths</th>
<th>SMR (95% CI)</th>
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<tbody>
<tr>
<td>1930-2</td>
<td>20-65</td>
<td>12 (221)</td>
<td>68 (147-240)</td>
</tr>
<tr>
<td>1949-53</td>
<td>20-64</td>
<td>6 (27-158)</td>
<td>7 (156-225)</td>
</tr>
<tr>
<td>1930-2</td>
<td>20-65</td>
<td>5 (29)</td>
<td>6 (24)</td>
</tr>
<tr>
<td>1949-53</td>
<td>20-64</td>
<td>4 (30)</td>
<td>110 (281)</td>
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<table>
<thead>
<tr>
<th>Years</th>
<th>Age Group</th>
<th>Observed Deaths</th>
<th>SMR (95% CI)</th>
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<tr>
<td>1930-2</td>
<td>20-65</td>
<td>5 (6)</td>
<td>110 (256)</td>
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</table>

* Occupation codes: 1930-2 ferrous moulders 151-2; brass moulders 150; ferrous foundry furnace men and labourers 140-2,149,153-4; brass and bronze furnace men and labourers 155-6; 1949-53 ferrous moulders 132; brass moulders 131; iron and steel foundry furnace men and iron foundry labourers 121,122/5,134-5 non-ferrous foundry furnace men and labourers 137-8.

Death from pneumonia in iron founders was twice that in brass founders, and that in iron miners and quarrymen was 50% greater than in lead miners; but there was no corresponding excess risk of mortality from bronchitis. Generally, the pattern is compatible with a specific hazard from iron, possibly superimposed on a second hazard from metal fume more generally.

### Conclusions

A growing body of evidence suggests that free iron in body fluids promotes bacterial growth. In theory, therefore, occupations that entail exposure to iron fume or dust could carry an increased risk of infection. The principal data that allow a test of this come from national statistics of occupational mortality, and are subject to several well documented biases. In particular, the derivation of SMRs depends on occupational codings for the numerator and denominator that come from different sources (death certificates and censuses respectively); whereas PMRs, although derived from a single source (death certificates), may mislead if overall mortalities in an occupation are unusually high or low. There are other limits to the data—for example, scope for exposure misclassification in the case of death certification is inaccurate or ill defined; changes over time in diagnostic practice and classification; and in some analyses small numbers of observed deaths and correspondingly wide confidence intervals. Nor should the potential for confounding by non-occupational factors be ignored. Our findings have been adjusted for age and social class, and we have argued that confounding by smoking is unlikely to explain them, but the increases in relative risk in a number of analyses are comparatively modest, and vulnerable to the effects of unrecognised confounders.

Set against these concerns are the consistency and coherence of effect, apparent through several periods of analysis and for several outcomes. Analyses of occupational mortality indicate that workers exposed to metal fume and dust have high death rates from pneumonia, influenza, and bronchitis, but not from other infections.

Risks have tended to be higher in men working with ferrous compared with other metals, but the difference is not clear cut. Thus, the possibility of a general hazard of metal fume—for example, from impairment of macrophage function—cannot be ruled out. Nevertheless, the specific role of iron merits further investigation. An early priority is to establish whether exposure to metal fume is associated with a higher incidence of respiratory infection as well as a higher mortality, and whether the associ—

### Other studies

Published cohort studies of mortality in workers exposed to metal fume or dust from iron ore shed little extra light on the topic. In general these have not looked specifically at infective risk, and have lacked sufficient statistical power to answer this particular question. Where excess mortality from respiratory disease has been described, often the data have not been broken down by cause, although in two large studies excesses of pneumonia were apparent, with SMRs of 185 and 167 among welders in the north east of England and in Seattle, United States. An association has also been described between work in ferrous foundries and death from chronic bronchitis. By contrast a large European study of 11000 welders failed to show an excess of pneumonia, or of bronchitis, emphysema, or asthma, and no excesses were found in a cohort of iron ore miners from Minnesota.

Morbidity from infective illness in metal workers has been studied even less than mortality. An analysis of hospital admissions among American shipyard workers during 1942–5 and a cross sectional survey of British shipyard employees indicated no relation between pneumonia and exposure to metal fume; but in a survey of certified sickness absence during a 10 year period, four of 36 welders had absences attributed to pneumonia compared with none of a similar number of controls. A number of cross sectional studies have failed to find an excess of chronic bronchitis in welders, but many studies have identified an excess of current cough and sputum in active workers, independent of smoking. Furthermore, a recent longitudinal study has described an enhanced deterioration in lung function in welders which was also independent of smoking.
ation is with all types of metal or only with iron. It would also help to know more about the fate of inhaled iron fume, and particularly what proportion remains unbound, for how long, and at what sites in the lung.

We thank Paul Winter and Hazel Inskip for their assistance in data analysis, and Christopher Martyn and Keith Godfrey for their constructive comments.

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